Predicting Double Replacement Reactions

Complete and Balance the following reactions. If no reaction occurs then write N.R.

1)
$$Na_2SO_4$$
 (aq) + $Ba(NO_3)_{2(aq)}$ ---->

2)
$$NaNO_{3(aq)} + NH_4Cl_{(aq)} ---->$$

3)
$$Pb(NO_3)_{2 (aq)} + Na_2CrO_{4(aq)}$$

4)
$$ZnCl_{2(aq)} + K_2CO_{3(aq)} --->$$

6) Barium Acetate + Copper II Chromate

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7) Solutions of Silver Nitrate and ammonium chloride are mixed
8) Solutions of Lead II nitrate and Sodium Chloride are mixed
9) Solutions of zinc sulfate and magnesium chloride are mixed
10) Solutions of Ammonium phosphate and Zinc Chlorate are mixed
11) Solutions of Aluminum bromide and Iron II Iodide are mixed

12) Solutions of sodium sulfide and Iron II Chlorate are mixed

13) Solutions of copper II bromide and potassium phosphate are mixed

14) Solutions of barium acetate and zinc sulfate are mixed