- 1. Volume and Density Lab Journal Entry:
 - a. Part 1
 - i. Objective: use displacement method and standard measurement method to determine the volume of an object
 - ii. Procedure:
 - 1. Record mass of copper plate to the nearest tenth
 - 2. Record the length, width, and height of the copper plate to the nearest tenth of a cm.
 - 3. Calculate the volume of the copper plate.
 - 4. Take the mass of the second object to the nearest tenth of a gram.
 - 5. Fill a 100 ml graduated cylinder between 70 80 ml of water. Record the exact level of the water to the nearest tenth of a ml.
 - 6. Place the object into the filled graduated cylinder and record the new level of water.
 - Subtract the old level from the new level to determine the amount of water displaced. (= to the volume of the object)
 - b. Part 2
 - i. Objective: calculate density and percent error
 - ii. Procedure:
 - 1. Use the information from part 1 to calculate the density of the copper and the object.
 - 2. Obtain the identity of your unknown object from the teacher and look up the density of that object and of copper in the MERK index.
 - 3. Record the information from the MERK index and calculate your % error for both the copper and the object.
 - 4. Create a data table of all your data from both part 1 and 2
 - 5. Under the data table describe some possible reasons for your % error other than just stating human error.